## Y7 Particles and Mixtures Homework Grids

Name: $\qquad$
Science Teacher:

|  | Comment |
| :--- | :--- |
| Grid 1.1 |  |
| Grid 1.2 |  |
| Grid 1.3 |  |
| Grid 1.4 |  |
| Grid 1.5 |  |
| Grid 1.6 |  |
| Grid 1.7 |  |
| Grid 1.8 |  |

## Grid 1.1: Use KO 13-18

## Due:

$\qquad$


The diagram to the left shows particle arrangements the three states.

Which arrangement shows a
Solid
Liquid
Gas
$\qquad$

Can you give at least 3 examples of
Solid $\qquad$
$\qquad$
Liquid $\qquad$
Change of state $B$ is called $\qquad$
Change of state C is called $\qquad$
Change of state $D$ is called $\qquad$
$\qquad$

## Grid 1.2: Use KO 13-18

## Due:

Fat and water in the sausages changed state.
Draw one line from each statement to the correct change of state.
Draw only two lines.
liquid to gas
fat melted
gas to liquid
liquid to solid
Susie uses charcoal as the fuel for the barbecue.
Which gas is needed forfuels to burn?
Draw a circle around the correct answer
Water
Oxygen
Carbon Dioxide
Nitrogen


$$
19
$$

正
water evaporated

## Grid 1.3: Use KO 13-18

Due: _

Kate made some ice cubes from pure water.
She used a sensor to measure the temperature of the ice.

What temperature will the sensor show when the ice is melting?
$\qquad$ ${ }^{\circ} \mathrm{C}$

Kate made some more ice cubes from salt solutions. She used a different amount of salt in each ice cube. The table below shows the temperature at which the ice cubes melted.

| mass of salt in <br> each ice cube $(\mathrm{g})$ | temperature ice <br> cube melted $\left({ }^{\circ} \mathbf{C}\right)$ |
| :---: | :---: |
| 5 | -4 |
| 10 | -8 |
| 15 | -11 |
| 20 | -15 |



Look at the table on the left.
As the mass of salt increased, what happened to the temperature at which the ice cube melted?
$\qquad$
$\qquad$

## Grid 1.4: Use KO 13-18

## Due:

$\square$

In very cold weather a mixture of salt and sand is spread on roads.

Why are salt and sand used?
Tick the correct two boxes


Salt makes the roads white.

Salt makes water freeze.


Sand increases friction between car tyres and the road.

Salt makes ice melt.


Sand dissolves in water.

Sand makes water freeze.

$\square$

## Grid 1.5: Use KO 13-18

## Due:

The table shows the melting points of four metals.

Which metal in the table has the highest melting point?

Which metal in the table has the lowest melting point?

Gold can be a gas or a liquid or a solid.
Choose from these words to fill the gaps below.

When gold is heated from room temperature to $1070^{\circ} \mathrm{C}$, the gold changes from a
to a $\qquad$ . .

What state (solid, liquid and gas) is Sodium metal at room temperature ( $25^{\circ} \mathrm{C}$ )?
$\qquad$

What state is iron at $1600^{\circ} \mathrm{C}$ ?
$\qquad$

## Grid 1.6: Use KO 13-18

Due:
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## Grid 1.7: Use KO 17

## Due:

Amy's family are at the beach during the summer. Amy and her sister have a bucket containing seawater and sand.

Seawater contains dissolved salt.
Describe what Amy can do to separate and collect pure water from seawater.

Seawater contains dissolved salt.
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Decide if the following statements are true or false

1. Water is a solvent for salt TRUE / FALSE
2. Sand sinks in water because water is more dense than sand TRUE / FALSE
3. When a solid dissolves in water, the solid is called a solute
TRUE / FALSE

Using a pencil, draw a labelled diagram to show the equipment used to separate sand from water.

## Grid 1.8: Use KO 13-18

## Due:

$\qquad$

Draw a line from each of the substances above to the group that it belongs to.
Draw only three lines.
Draw a line from each group to the correct description.
Draw only three lines
substance

group


It contains only one type of atom.


Two or more types of atoms are chemically joined together.

